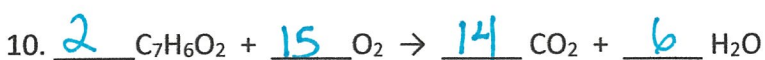
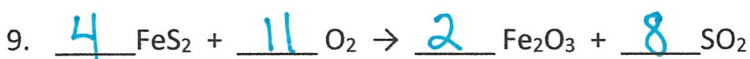
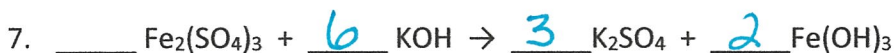
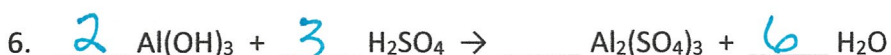
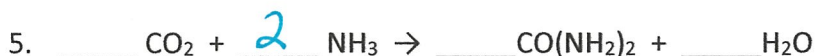
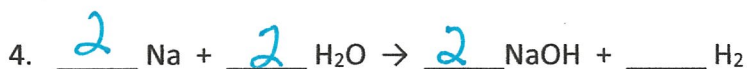
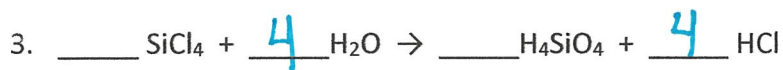


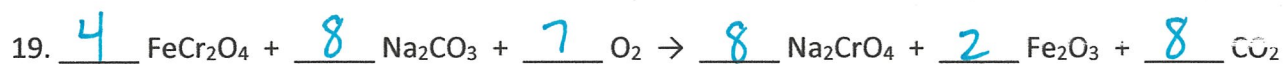
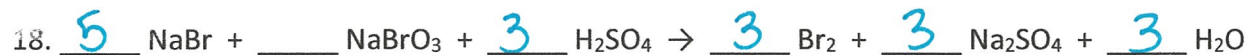
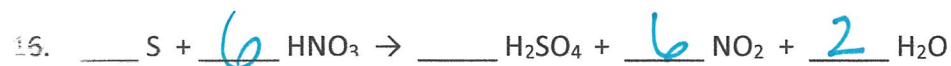
Balancing Chemical Equations Practice

Key

11. Dicarbon dihydride + Oxygen \rightarrow Carbon dioxide + Water12. Potassium oxide + Water \rightarrow Potassium Hydroxide13. Aluminum + Oxygen \rightarrow Aluminum oxide14. Iron (III) chloride + Ammonium hydroxide \rightarrow Iron (III) hydroxide + Ammonium15. Calcium chlorate \rightarrow Calcium chloride + oxygen

BONUS CHALLENGE QUESTIONS:

Key



**Note: Although Noble gases such as Xenon are normally unreactive, they can sometimes form compounds. Xenon tetrafluoride was the first discovered binary compound of a Noble gas and it can be formed under very high temperatures.*